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SYNTHESIS OF THE INDOLE ALKALOID ELLIPTICINE.

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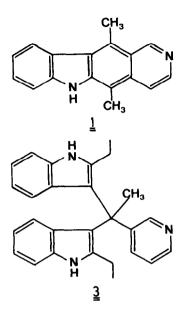
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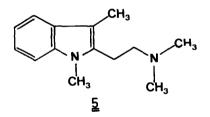
The potentially useful anti-cancer activity of ellipticine (<u>1</u>) and some of its derivatives, has triggered considerable synthetic efforts¹⁻⁶ in this area. In spite of the fact that no less than 20 routes are available at present no convenient method suitable for large-scale preparations does exist. Actually the inaccessability⁷ of ellipticine derivatives has hampered the development of biological evaluation.

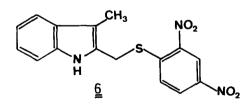
Some years ago we showed⁸ that 2-ethylindole⁹ and 3-acetylpyridine¹⁰ could be condensed in acetic acid (or methanolic HBr) to give <u>2</u> rather than the expected compound <u>3</u> (<u>cf</u>. ref 11). At the time the idea was to cyclize <u>2</u> or <u>4a</u> by introducing a suitable strongly electron-withdrawing substituent such as CH₃CO-, $C_{6}H_{5}CO$ - or CH₃SO₂-, on the nitrogen atom in the pyridine ring, which would induce desired electrophilicity in the 4-position of the pyridine ring. This electrophilic centre should then attack the enamine system in the exocyclic tautomer of <u>e.g. 4a</u> (<u>i</u>. e. 4b) to give the desired cyclization.

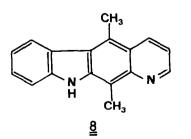
A few examples of tautomerization of 2-alkylindoles followed by electrophilic attack are recorded in the literature. Thus, Thesing and Semler have shown¹² that 1,2,3-trimethylindole, on treatment with formaldehyde and dimethylamine, yielded the Mannichs base 5. More recently, Hutzinger and Raj found¹³ that 2,3-dimethylindole when treated with 2,4-dinitrosulfenylchloride gave 3-methylindole-2-methylene-2',4'-dinitrophenyl sulfide (<u>6</u>). The fact that the 2-CH₃ group in 2,3-dimethylindole is completely deuteriated under mild acidic conditions¹⁴ gave further support to the suggested approach.

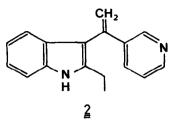
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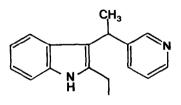




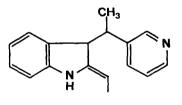




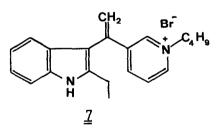


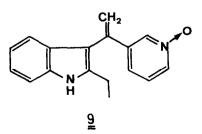






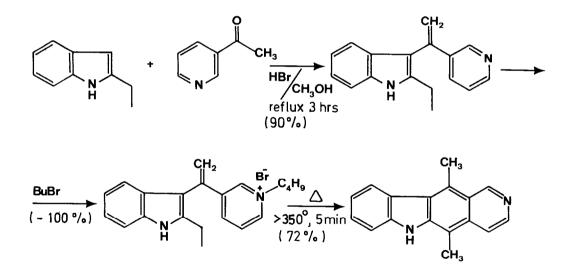
<u>4b</u>





In spite of considerable efforts to effect the desired cyclization in various solvents in the temperature range 0-100[°] the yields were invariably low (0-0.5 %). However, pyrolysis (5 min, > 350°) of N-alkylated derivatives, e.g. 7 gave ellipticine in high yield (72 %). The desired product could readily be separated from the minor products, which included 2-ethylindole, 2, and an isomer assigned structure $\underline{8}^{15}$. It was also found that pyrolysis of $\underline{9}^{16}$ (and even 2) likewise afforded ellipticine (<u>albeit</u> in relatively low yields). The heating technique during the pyrolysis is essential. Thus rapid heating of 7 gave good yields of ellipticine (and less than 10 % of $\underline{8}$), whereas slow heating ($\underline{e.g.} 220^{\circ}$ for 30 min) gave $\underline{8}$ as the predominant product. Similar results were obtained by heating ($\underline{80^{\circ}}$) 7 with NaOC₂H₅ in ethanol.

To sum up ellipticine (and several derivatives 17) can very efficiently be synthesized by the following route



The possibility to integrate all the steps to a one-pot synthesis has not been explored but does indeed look promising.

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- 16. (a) Compound <u>9</u>, m.p. 157-8^o, was prepared by condensation (<u>cf</u>. ref 8) of 2-ethylindole and 3-acetylpyridine-<u>N</u>-oxide^{16b}. Oxidation of <u>2</u> did not give <u>9</u>.
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